

KLEA® 407C



Physical Property Data Sheet SI Units

PHYSICAL PROPERTY DATA FOR KLEA®407C			
PROPERTY		UNITS	VALUE
Bubble Point	(1 atm)	°C	-44.0
Dew Point	(1atm)	°C	-36.8
Bubble Point Pressure	(25°C)	bara	11.9
Estimated Critical Temperature		°C	86
Latent Heat of Vaporisation	(Tm=25°C)	kJ/kg	193.9
Trouton's constant		kJ/kg.K	1.08
Coeff.Vol.Therm.Exp.	(LIQ,0-20°C)	K ⁻¹	0.0037
Density (SAT VAPOUR) at 1 atm		kg/m ³	4.57

EQUATION OF STATE (MARTIN-HOU)

$$P_r = \frac{X T_r}{V_r - B} + \sum_{i=1, 4} \frac{(A_i + B_i T_r + C_i \exp(-K T_r))}{(V_r - B)^{(i+1)}}$$

Where :

$$T_r = T/T_c, P_r = P/P_c, V_r = V/V_c = V \times \text{RHO}_c$$

$$X = 3.649216$$

$$B = 0.0$$

$$K = 5.475$$

$$T_c, P_c, \text{RHO}_c = 359.2(\text{K}), 46.52(\text{bara}), 490(\text{kg/m}^3)$$

$$A_1, B_1, C_1 = -11.576411672, 6.9516394791, -12.992106787$$

$$A_2, B_2, C_2 = 9.108927853, -5.8551996466, -101.85219827$$

$$A_3, B_3, C_3 = -4.2674117378, 0.0, 0.0$$

$$A_4, B_4, C_4 = 17.976354749, 0.0, -1992.5643203$$

Applicable range:- 0-30 bara, 0-100K superheat.

SATURATION ENVELOPE – BUBBLE POINT TEMPERATURES	
Bubble Point Temperature (T_b) = $A + BX + CX^2 + DX^3$	
A = 228.9073	T_b = Bubble point temperature in K
B = 20.99838	X = Ln(P)
C = 1.855389	P = pressure in bara
D = 0.37783	

The correlations in this data sheet should not be used outside the applicable ranges quoted.
Contact Mexichem Fluor for further advice.

SATURATION ENVELOPE – DEW POINT TEMPERATURES	
$T_d = A + BX + CX^2 + DX^3$	
A = 236.0528 B = 20.37606 C = 2.17732 D = 0.250429	T_d = Dew point temperature in K X = Ln(P) P = pressure in bara

SATURATION ENVELOPE – MID POINT TEMPERATURES	
$T_m = A + BX + CX^2 + DX^3$	
A = 232.4902 B = 20.64264 C = 2.047962 D = 0.3082967	T_m = Average of Dew & Bubble point temperatures in K X = Ln(P) P = pressure in bara

LATENT HEAT OF VAPORISATION	
$DH_{vap} = A + Bx + Cx^2 + Dx^3 + Ex^4$ Where $x = (1 - (T_m/T_c))^{(1/3)}$	
A = 218.8494 B = -1683.2244 C = 6325.3877 D = -8231.939 E = 3884.46	T_m = Mid point temperature K T_c = Critical Temperature K $T_c = 359.2K$ $DH_{vap} = kJ/kg$

IDEAL GAS HEAT CAPACITY	
$C_p(\text{ideal}) = A + BT + CT^2 + D/T$	
A = -0.083843 B = 0.00321516 C = -1.94062E-06 D = 36.35655	T = Temperature K C_p (ideal) kJ/kg.K

The correlations in this data sheet should not be used outside the applicable ranges quoted. Contact Mexichem Fluor for further advice.

SATURATED LIQUID ENTHALPY

$$H_{liq} = A + Bx + Cx^2 + Dx^3 + Ex^4$$

where $x = (1 - (T_b/T_c))^{(1/3)}$

A = 131.7716
B = 1224.77
C = -4212.446
D = 5045.1332
E = -2528.57

T_b = Bubble point temperature K
 T_c = Critical Temperature K
 $T_c = 359.2$ K
 H_{liq} kJ/kg

LIQUID DENSITY

$$d_{liq} = A + Bx + Cx^2 + Dx^3 + Ex^4$$

where $x = (1 - (T_b/T_c))^{(1/3)}$

A = -650.4582
B = 7861.656
C = -15312.36
D = 15705.71
E = -5734.191

T_b = Bubble point temperature K
 T_c = Critical Temperature K
 $T_c = 359.2$ K
 d_{liq} kg/m³

LIQUID VISCOSITY

$$\ln(\mu_{liq}) = A + B/T_m + CT_m + DT_m^2$$

A = 15.66442
B = -1283.053
C = -0.061504
D = 5.81907E-5

T_m = Mid point temperature K
 μ_{liq} cP

LIQUID THERMAL CONDUCTIVITY

$$K_{liq} = A + BT_m + CT_m^2 + DT_m^3$$

A = 0.11898
B = 0.0
C = -6.71955E-7
D = 9.039943

T = Mid point temperature K
 K_{liq} W/m K

The correlations in this data sheet should not be used outside the applicable ranges quoted.
Contact Mexichem Fluor for further advice.

SATURATED VAPOUR DENSITY

$$d_{\text{vap}} = A + Bx + Cx^2 + Dx^3 + Ex^4$$

where $x = (1 - (T_d/T_c))^{(1/3)}$

A = 315.4088
B = 343.5035
C = -3883.4765
D = 5587.518
E = -2353.113

T_d = Dew point temperature K
 T_c = Critical Temperature K
 T_c = 359.2 K
 d_{vap} kg/m³

VAPOUR VISCOSITY (IDEAL VAPOUR)

$$\mu_{\text{vap}} = A + BT + CT^2$$

A = -0.0013724
B = 5.359E-5
C = -1.53122E-8

T_m = Temperature K
 μ_{vap} cP

VAPOUR VISCOSITY (SATURATED VAPOUR)

$$\mu_{\text{vap}} = A + BT_d + CT_d^2 + D/T_d$$

A = 0.375554
B = -1.36789E-3
C = 1.77845E-6
D = -33.3749

T_d = Dew point temperature K
 μ_{vap} cP

VAPOUR THERMAL CONDUCTIVITY (IDEAL VAPOUR)

$$K_{\text{gas}} = A + BT + CT^2$$

A = -3.5565E-3
B = 3.5848E-5
C = 7.74837E-8

T_d = Temperature K
 K_{gas} W/m.K

The correlations in this data sheet should not be used outside the applicable ranges quoted. Contact Mexichem Fluor for further advice.

SATURATED VAPOUR THERMAL CONDUCTIVITY	
$K_{\text{gas}} = A + BT_d + CT_d^2 + D/T_d$	
A = 0.42538 B = -1.56445E-3 C = 2.08498E-6 D = -38.54645	T_d = Dew point temperature K K_{gas} W/m.K

SPEED OF SOUND (SATURATED VAPOUR)	
$\mu_{\text{vap}} = A + BT_d + CT_d^2 + D/T_d$	
A = -869.9 B = 4.927595 C = -7.5007E-3 D = 68779.76	T_d = Dew point temperature K μ_{vap} = m/s

SATURATION ENVELOPE			
PRESSURE Bara	TEMPERATURES (°C)		
	BUB pnt	MID pnt	DEW pnt
1.0	-44.2	-40.7	-37.1
2.0	-28.7	-25.3	-21.8
3.0	-18.4	-15.1	-11.8
4.0	-10.6	-7.3	-4.0
6.0	1.5	4.7	7.8
8.0	10.8	13.9	16.9
10.0	18.6	21.5	24.4
12.0	25.2	28.0	30.8
15.0	33.7	36.4	39.0
20.0	45.5	47.8	50.2
25.0	55.2	57.3	59.4
30.0	63.5	65.4	67.2

The correlations in this data sheet should not be used outside the applicable ranges quoted. Contact Mexichem Fluor for further advice.

LIQUID PROPERTIES					
TEMP	LIQUID DENSITY	LIQUID ENTH	LATENT HEAT	LIQUID VISCOSITY	LIQ.THERM COND
°C	kg/m ³	kJ/kg	kJ/kg	cP	W/m.K
-50.0	1399	30.7	258.9	0.40	0.126
-40.0	1369	44.5	251.3	0.36	0.121
-30.0	1337	58.2	243.7	0.32	0.116
-20.0	1305	72.0	235.9	0.29	0.112
-10.0	1271	85.9	227.8	0.25	0.107
0.0	1235	100.0	219.2	0.23	0.102
10.0	1197	114.0	209.9	0.20	0.097
20.0	1157	129.0	199.6	0.18	0.092
25.0	1135	137.0	193.9	0.16	0.090
30.0	1113	145.0	187.8	0.15	0.087
40.0	1063	162.0	174.1	0.14	0.082
50.0	1005	180.0	157.7	0.12	0.077

For Liquid Density and Liquid Enthalpy the bubble point temperatures are used. For Latent Heat, Liquid Viscosity and Liquid Thermal Conductivity mid point temperatures are used.

IDEAL GAS PROPERTIES			
TEMP	ID. GAS Cp	ID. GAS VISCOSITY	ID. GAS THERM COND
°C	kJ/kg.K	cP	W/m.K
-50.0	0.700	0.0098	0.0083
-40.0	0.716	0.0103	0.0090
-30.0	0.733	0.0108	0.0097
-20.0	0.749	0.0112	0.0105
-10.0	0.766	0.0117	0.0112
0.0	0.783	0.0121	0.0120
10.0	0.799	0.0126	0.0128
20.0	0.816	0.0130	0.0136
25.0	0.824	0.0132	0.0140
30.0	0.832	0.0135	0.0144
40.0	0.849	0.0139	0.0153
50.0	0.865	0.0143	0.0161

The correlations in this data sheet should not be used outside the applicable ranges quoted. Contact Mexichem Fluor for further advice.

SATURATED VAPOUR PROPERTIES				
TEMP	SAT VAP	SAT VAP	SAT VAP	SPEED
°C	DENSITY	VISCOSITY	THERM COND	OF SOUND
	kg/m ³	cP	W/m.K	m/s
-50.0	-	0.0093	0.0074	164.0
-40.0	3.94	0.0102	0.0086	166.0
-30.0	6.17	0.0108	0.0097	168.0
-20.0	9.31	0.0114	0.0107	169.0
-10.0	13.57	0.0119	0.0116	169.0
0.0	19.24	0.0124	0.0125	168.0
10.0	26.67	0.0130	0.0134	167.0
20.0	36.35	0.0135	0.0144	165.0
25.0	42.24	0.0139	0.0150	163.0
30.0	48.95	0.0142	0.0156	161.0
40.0	65.45	0.0150	0.0168	157.0
50.0	87.38	0.0160	0.0183	152.0

The temperatures used are Dew Point Temperatures.

The correlations in this data sheet should not be used outside the applicable ranges quoted. Contact Mexichem Fluor for further advice.

KLEA® 407C



Mexichem.
FLUOR

**Mexichem UK Limited, The Heath Business and Technical Park,
Runcorn, Cheshire, WA7 4QX
Telephone: +44 (0) 1928 514840
www.mexichemfluor.com**

Information contained in this publication, or as otherwise supplied to the Users is believed to be accurate and given in good faith, but it is for the User to satisfy itself of the suitability for its own particular purpose. Mexichem Fluor gives no warranty as to the fitness of the Product for any particular purpose and any implied warranty or condition (statutory or otherwise) is excluded except to the extent that such exclusion is prevented by law. Mexichem Fluor accepts no liability for loss or damage (other than that arising from death or personal injury caused by defective product, if proved), resulting from reliance on this information. Freedom under Patent, Copyright and Design cannot be assumed. KLEA® and Mexichem Fluor® are trademarks of Mexichem SAB de C.V.
© Mexichem Fluor 2010. All rights reserved. Not to be reproduced without the consent of the copyright owner.